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DOI:

[10.1080/17436753.2017.1366733](https://doi.org/10.1080/17436753.2017.1366733)

Document Version

Accepted author manuscript

[Link to publication record in Manchester Research Explorer](#)

Citation for published version (APA):

Hall, D., & Tangsritrakul, J. (2018). Structural and functional characterisation of KNNS-BNKZ lead-free piezoceramics. *Advances in Applied Ceramics*, [dx.doi.org/10.1080/17436753.2017.1366733]. <https://doi.org/10.1080/17436753.2017.1366733>

Published in:

Advances in Applied Ceramics

Citing this paper

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Structural and functional characterisation of KNNS-BNKZ lead-free piezoceramics

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Abstract

Promising piezoelectric properties have been reported recently for lead-free $0.96(\text{K}_{0.48}\text{Na}_{0.52}\text{Nb}_{0.95}\text{Sb}_{0.05})-0.04\text{Bi}_{0.5}(\text{Na}_{0.82}\text{K}_{0.18})_{0.5}\text{ZrO}_3$ (KNNS-BNKZ) ceramics. The presence of coexisting ferroelectric rhombohedral and tetragonal phases is thought to play a key role in their functional properties, but a thorough understanding is currently lacking. In this experiment, $(1-x)\text{KNNS}-(x)\text{BNKZ}$ ceramics with $x = 0$ to 0.05 were prepared by the mixed-oxide method. High resolution synchrotron x-ray powder diffraction (SXPd) measurements reveal that the addition of BNKZ into KNNS ceramics leads to an increase of the rhombohedral-orthorhombic phase transition temperature ($T_{\text{R-O}}$) and a reduction of the orthorhombic-tetragonal phase transition temperature ($T_{\text{O-T}}$) leading to orthorhombic-tetragonal and rhombohedral-tetragonal phase coexistence at room temperature for compositions with $x = 0.02$ and 0.04 , respectively. By combining the results of the SXPd measurements with microstructural examination using SEM, evidence is also found for the occurrence of chemical heterogeneity, which could provide an additional means to control the functional properties. The structural observations are correlated with changes in the dielectric properties, obtained as permittivity-temperature plots, and variations in the polarisation and coercive field values, obtained from measurements of the ferroelectric hysteresis loops.

Keywords: Phase coexistence; Lead-free; Ferroelectric; Synchrotron x-ray diffraction.

1. Introduction

Piezoelectric ceramics are widely used in many applications as electromechanical sensors, actuators and transducers in applications such as fuel injectors, solid state motors and printers [1, 2]. In lead zirconate titanate (PZT), it is well known that compositions near the morphotropic phase boundary (MPB) give rise to materials having excellent piezoelectric properties because of rhombohedral-tetragonal (R-T) phase coexistence, which plays an important role in its electrical properties. However, lead is a toxic element that can cause health and environmental problems [1, 3, 4, 5].

Potassium sodium niobate ($K_xNa_{1-x}NbO_3$ or KNN) has been considered as one of the leading lead-free piezoceramics since 2004, when Saito et al. reported high piezoelectric coefficients, comparable to those of some PZT ceramics, for a Li, Ta and Sb modified KNN.[6] It should be possible to replace the commercial PZT devices with lead-free piezoceramics in the future by improving the piezoelectric properties and their temperature stability[7]. There are many studies that focus on the study of two phase coexistence in order to enhance piezoelectric properties of lead-free piezoceramics [8, 9, 10, 11]. The control of polymorphic phase transitions (PPT) has been used to form coexisting phase at room temperature to improve the electrical properties [1]. It has been found that the substitution of $(Bi_{0.5}Na_{0.5})^{2+}$, $[Bi_{0.5}(Na_{0.7}K_{0.2}Li_{0.1})_{0.5}]^{2+}$ or $[Bi_{0.5}(Na_{1-w}K_w)_{0.5}]^{2+}$ species can reduce T_{O-T} (orthorhombic-tetragonal transition temperature) while Zr^{4+} ions can increase T_{R-O} (rhombohedral-orthorhombic transition temperature). As a result, the O-T, R-O-T and R-T phase coexistence can be observed near room temperature [8, 12, 13].

Wang et al. [8] reported that $0.96(\text{K}_{0.48}\text{Na}_{0.52}\text{Nb}_{0.95}\text{Sb}_{0.05})-0.04\text{Bi}_{0.5}(\text{Na}_{0.82}\text{K}_{0.18})_{0.5}\text{ZrO}_3$ ceramic, which was said to contain R-T mixed phases, exhibited a d_{33} of 490 pC/N. The promising piezoelectric properties of such materials merit further investigations to clarify the relationships between their complex multi-phase structure and functional properties. The aim of this experiment is to study the effects of adding $\text{Bi}_{0.5}(\text{Na}_{0.82}\text{K}_{0.18})_{0.5}\text{ZrO}_3$ (BNKZ) into $\text{K}_{0.48}\text{Na}_{0.52}\text{Nb}_{0.95}\text{Sb}_{0.05}\text{O}_3$ (KNNS) on the phase formation using high resolution synchrotron x-ray powder diffraction (SXPd). Moreover, the effects of BNKZ on the microstructures, dielectric properties and ferroelectric properties are also investigated.

2. Experimental procedures

Compositions were prepared by the conventional mixed-oxide method according to the chemical formula $(1-x)(\text{K}_{0.48}\text{Na}_{0.52}\text{Nb}_{0.95}\text{Sb}_{0.05})-(x)(\text{Bi}_{0.5}(\text{Na}_{0.82}\text{K}_{0.18})_{0.5}\text{ZrO}_3)$ with $x = 0$ to 0.05, denoted as 0BNKZ to 5BNKZ. The raw powders of K_2CO_3 (99.0%), Na_2CO_3 (99.8%), Nb_2O_5 (99.0%), Sb_2O_3 (98.0%), Bi_2O_3 (99.0%) and ZrO_2 (99.0%) were weighed according to the nominal composition. However, it is well known that carbonate powders are moisture sensitive [14]. Therefore, in this experiment, K_2CO_3 and Na_2CO_3 powders were dried overnight in an oven at 90°C before weighing in order to avoid non-stoichiometry of the calcined powders. Then, the raw powders were milled in isopropanol for 24 hours and calcined at 850°C for 6 hours.

After calcination, the calcined powders were milled again for 24 hours and dried overnight before mixing with 2 wt% of PEG as a binder. Next, the calcined powder were pressed under a pressure of 25 MPa in a 8 mm diameter cylindrical steel die into a disc with thickness around 1 mm. The green pellets were covered in a calcined powder

of the same composition to avoid volatilisation of alkali oxides and sintered at 1120°C to 1170°C for 3 hours after binder burn out.

The density of the samples was measured by the Archimedes method with H₂O as the immersion liquid, using the average value from three samples for each composition. The theoretical densities were calculated on the basis of the refined crystallographic parameters together with the nominal chemical composition. Crystallographic information was obtained using high resolution synchrotron x-ray powder diffraction (SXPd) on beamline I11 at the Diamond Light Source, UK using a wavelength of 0.826 Å. The KNNS-BNKZ ceramics were crushed into powder and compacted into borosilicate glass capillaries with a diameter of 0.3 mm. A high resolution MAC detector, consisting of 5 banks of 9 Si single crystal detectors, was used to scan the angular range from 10° to 90° 2θ at room temperature for 20 minutes. The full-pattern refinement was accomplished by a Rietveld refinement method using Topas, version 4.2.

The microstructure of polished cross-sections was examined using a Philips XL30 FEGSEM in backscattered electron mode. For electrical measurements, the ceramic discs were ground to around 1 mm in thickness before applying and firing a silver electrode (Gwent Group type C2000107P3) at a temperature of 550°C for 30 minutes. The temperature dependence of dielectric permittivity of KNNS-BNKZ ceramics was measured at frequencies between 1 and 100 kHz using a HP4284A impedance analyser over the temperature range 50°C to 450°C. The ferroelectric P-E hysteresis loops of KNNS-BNKZ ceramics were measured at room temperature. The samples were measured in a silicone oil bath to avoid electrical breakdown. Four cycles of a sinusoidal electric field with amplitude 4 kV mm⁻¹ were applied at a frequency of 2 Hz to test the samples. The voltage-time waveforms were generated using a HP33120A

function generator in combination with a Chevin Research HVAIB high voltage amplifier, while the induced current was measured by means of a current amplifier. The applied voltage and current waveforms were measured using a 16-bit A/D card (Measurement Computing USB-1608FS) and the P-E hysteresis loops constructed using LabVIEW software.[15]

3. Results and discussion

3.1 Density

The relative densities of the 1BNKZ to 5BNKZ ceramics were higher than 93%, while that of 0BNKZ was lower than 90%, as shown in Table 1. This is possibly due to the incorporation of bismuth oxide, Bi_2O_3 , since it was reported by Du [16] that the addition of Bi_2O_3 to KNN acts to increase the melting point and optimum sintering temperature of the solid solution. Therefore, the higher sintering temperature used with bismuth-containing compositions enhances the densification. Consequently, the composition 0BNKZ in our experiment, in the absence of bismuth oxide, exhibited poor density and many large pores, as illustrated by the micrograph presented in figure 4(a). Previous studies on Sb-doped KNN ceramics have employed other sintering aids such as CuO in their sample preparation [17, 18, 19].

3.2 Dielectric properties

Figure 1 illustrates the temperature-dependence of dielectric properties for KNNS-BNKZ ceramics, measured at frequencies from 1 to 100 kHz. The Curie temperature (T_C) of 0BNKZ occurred at 302°C, while T_{O-T} was approximately 125°C. The high dielectric losses of the 0BNKZ ceramic, which are most prominent at high temperatures and low frequencies, indicate a relatively high conductivity associated

with the high porosity. It is evident that the addition of BNKZ into KNNS causes shifts in the transition temperatures, reducing T_C from 302°C to 214°C and T_{O-T} from 125°C to lower than 50°C. These results are consistent with those of Wang [8], who reported that the substitution of $[\text{Bi}_{0.5}(\text{Na}_{0.82}\text{K}_{0.18})_{0.5}]^{2+}$ in KNN reduces T_{O-T} as noted above. The influence of Zr^{4+} on T_{R-O} could not be verified in the present results. Furthermore, the addition of BNKZ gave rise to a relaxor ferroelectric character in the ϵ_r -T relationships, giving rise to a broadening of the Curie peak and frequency-dependence of the permittivity at temperatures below T_C . The phase transformation became very diffuse for the 5BNKZ ceramic in comparison with the other compositions, as shown in figure 1(d). Xing [20] also observed a diffuse phase transition in KNN-based ceramics having compositions $(1-x)[\text{K}_{0.48}\text{Na}_{0.52}\text{Nb}_{0.95}\text{Sb}_{0.05}\text{O}_3]-(x)[(\text{Bi}_{0.5}\text{Na}_{0.5})_{0.9}(\text{Li}_{0.5}\text{Ce}_{0.5})_{0.1}\text{ZrO}_3]$ when x exceeded 0.045, which was attributed to reaching the solubility limit. In terms of dielectric loss, relatively high $\tan\delta$ values were generally observed at temperatures below T_C due to the presence of ferroelectric domains [21]. With increasing temperature, the losses reduced in the region of T_C , but then increased dramatically, particularly when measured at low frequencies. The frequency-dependence of $\tan\delta$ in the high temperature region is characteristic of a conduction mechanism, as described by Hardtl [21].

Figure 1. Temperature-dependence of dielectric properties for (a) 0BNKZ, (b) 2BNKZ, (c) 4BNKZ and (d) 5BNKZ.

3.3 Identification of crystal structures at room temperature

The high resolution SXPD results for KNNS-BNKZ ceramics, which illustrate the effects of introducing BNKZ on the diffraction peak profiles, are presented figure 2. It is evident that 0BNKZ had a predominantly orthorhombic structure, indicated by characteristic splitting of the pseudo-cubic $\{200\}_p$, $\{210\}_p$ and $\{211\}_p$ reflections, for example. The addition of BNKZ had a dramatic effect on the SXPD peak profiles at room temperature, which changed in intensity and became broader, suggesting the presence of overlapping peaks due to the coexistence of different phases. A secondary phase, tentatively identified as $K_{2.75}Nb_{5.45}O_{15}$, was evident in the SXPD patterns of specimens with higher BNKZ contents, as indicated in figure 2(a).

Figure 2. The SXPD peak profiles of 0BNKZ, 2BNKZ and 4BNKZ ceramics. (a) Representative diffraction patterns with pseudo-cubic indices and (b) enlargement of $(200)_p$ peak profiles.

The single-phase orthorhombic structure for 0BNKZ was confirmed by full-pattern refinement, giving the results shown in figure 3(a) and Table 1. For 2BNKZ, the best fit was obtained by a 2-phase O-T mixture with a T content of approximately 37%, as illustrated in figure 3(b). According to the dielectric property measurements shown in figure 1(b) above, the O-T transition temperature, T_{O-T} , of 2BNKZ was approximately 100°C, which is substantially higher than room temperature. Therefore, it is evident that the incorporation of BNKZ into KNN led to a significant broadening of the phase transition region since a significant proportion of T phase was still present at room temperature.

When the BNKZ content increased further, the orthorhombic phase fraction decreased and disappeared due to the combined effects of an increase of T_{R-O} and reduction of T_{O-T} as well as T_C [8]. As a result, an R-T mixed phase containing approximately 56% R was identified in the composition 4BNKZ. These results are consistent with those of Wang et al. [8], who inferred the presence of an R-T mixed

phase from their temperature-dependent dielectric measurements. However, there are some peaks of 4BNKZ that are not well fit, as observed in both (220)_p and (222)_p reflections in figure 3(c). This is possibly a consequence of the chemical heterogeneity, as described below in section 3.4. In this study, we have been able to identify the phase coexistence in this material directly using high resolution SXPD. The volume fractions of the different coexisting phases as well as their crystallographic parameters and goodness-of-fit (χ^2) obtained by Rietveld refinement are summarised in Table 1. Note that the χ^2 in Table 1 is defined as the weight-profile R-factor (R_{wp}) divided by the expected R-factor (R_{exp}) [22].

Table 1. Coexisting phases, phase fractions, lattice parameters, χ^2 , R_{wp} and relative densities of 0BNKZ to 5BNKZ ceramics. Uncertainties in lattice parameters are less than 0.0001 Å.

Composition	Phases present	Phase fraction / %	Lattice parameter				χ^2	R_{wp}	Relative density / %
			a / Å	b / Å	c / Å	α / °			
0BNKZ	O	O = 100	a _O = 5.6272	b _O = 3.9506	c _O = 5.6495	-	5.86	12.1	87
1BNKZ	O-T	O = 75 T = 25	a _O = 5.6227 a _T = 3.9684	b _O = 3.9513 -	c _O = 5.6413 c _T = 3.9736	- -	2.96	7.7	97
2BNKZ	O-T	O = 63 T = 37	a _O = 5.6246 a _T = 3.9713	b _O = 3.9597 -	c _O = 5.6445 c _T = 3.9982	- -	3.91	10.4	93
3BNKZ	R-T	R = 20 T = 80	a _R = 3.9841 a _T = 3.9708	- -	- c _T = 3.9932	α = 89.72 -	4.59	12.3	96
4BNKZ	R-T	R = 45 T = 55	a _R = 3.9793 a _T = 3.9681	- -	- c _T = 3.9976	α = 89.83 -	5.48	14.9	96
5BNKZ	R-C	R = 46 C = 54	a _R = 3.9779 a _C = 3.9756	- -	- -	α = 89.83 -	5.77	15.9	99

Figure 3. The measured, calculated and difference data obtained by full-pattern refinement showing $(200)_p$, $(211)_p$, $(220)_p$ and $(222)_p$ reflections of (a) single O phase in 0BNKZ, (b) O-T coexisting phases in 2BNKZ and (c) R-T coexisting phases in 4BNKZ.

3.4 Microstructural observations

The cross-sectional microstructures of 0BNKZ and 2BNKZ ceramics were investigated in order to study on the effect of BNKZ on the grain size, illustrated in figure 4. It was found that the abnormal grain growth was observed in both 0BNKZ and 2BNKZ, with average grain sizes around $15.7 \pm 3.3 \mu\text{m}$ and $1.3 \pm 0.5 \mu\text{m}$, respectively. Malic [23] reported that secondary phases can possibly lead to the occurrence of abnormal grain growth in KNN-based ceramics.

Examination of the BNKZ-modified KNN ceramics revealed the occurrence of core-shell type microstructures, as shown in figure 4(b) for 2BNKZ. A second phase was also observed as lighter regions in the BSE images; this was identified as $\text{K}_{2.75}\text{Nb}_{5.45}\text{O}_{15}$ by SXPD, as noted above in section 3.3 and figure 2(a). Similar microstructures were observed in all other compositions except 0BNKZ, suggesting the presence of chemical heterogeneity [24, 25, 26]. The formation of core-shell type microstructures has been explained by exceeding the solubility limit of dopants, raw material heterogeneity or strain gradients that lead to controlled or suppressed diffusion processes [24]. In this study, it was found that it was not possible to eliminate the core-shell structures by using a longer sintering time of up to 6 hours.

Figure 4. Polished cross-sections of KNNS-BNKZ ceramics observed by BSE imaging (a) abnormal grain growth in 0BNKZ and (b) core-shell microstructure in 2BNKZ with lighter grains of $\text{K}_{2.75}\text{Nb}_{5.45}\text{O}_{15}$ secondary phase.

Additional high temperature SXPD measurements were undertaken to provide further evidence for this type of chemical segregation. It is anticipated that a chemically-homogeneous KNN ceramic should transform to a cubic structure, with no splitting of the diffraction peaks, at temperatures above the Curie point. In the present case, evidence was found for a persistent shoulder on all of the diffraction peaks for 2BNKZ measured at 337°C, well above its Curie temperature of ~276°C, as shown by the results presented in figure 5 for the 2BNKZ composition. A similar feature was also observed in the diffraction patterns for all of the compositions containing BNKZ, becoming more pronounced with increasing BNKZ content. This is attributed to the occurrence of a core-shell type microstructure comprising two cubic phases having different average chemical compositions and lattice parameters, which were determined as $a_{\text{core}} = 3.9793 \text{ \AA}$ and $a_{\text{shell}} = 3.9885 \text{ \AA}$. The lattice strain between these phases is approximately 0.23 %. Further investigation of the local variations in crystal symmetry and chemical composition using high resolution TEM would help to clarify the nature and origin of the core-shell microstructure.

Figure 5. Full-pattern refinement of 2BNKZ ceramic at 337°C showing $(200)_p$, $(211)_p$, $(220)_p$ and $(222)_p$ reflections of coexisting cubic phases with two lattice parameters due to core-shell microstructure.

3.5 Ferroelectric properties

The ferroelectric polarisation-electric field (P-E) hysteresis loops obtained for the KNNS-BNKZ ceramics are illustrated in figure 6. With increasing BNKZ content from 1BNKZ to 5BNKZ, the saturation polarisation (P_s), remnant polarisation (P_r) and coercive field (E_c) decreased from 0.24 to 0.12 C/m², 0.18 to 0.03 C/m², and 1.69 to 0.82 kV/mm, respectively. A strongly rounded loop was obtained for 0BNKZ, which was attributed to the effects of a relatively high electrical conductivity, consistent with the dielectric results presented in figure 1 above. These results show that the addition of the relaxor ferroelectric BNKZ could enhance the domain switching in KNN, giving rise to a reduction in E_c but with an associated reduction in P_r . For comparison, Wu [9] found the highest saturation polarisation of around 0.2 C/m² for the composition

$0.96\text{K}_{0.46}\text{Na}_{0.54}\text{Nb}_{0.95}\text{Sb}_{0.05}\text{O}_3\text{-}0.04\text{Bi}_{0.5}(\text{Na}_{0.82}\text{K}_{0.18})_{0.5}\text{ZrO}_3$, which is slightly higher than that observed in the present study.

It was anticipated that the addition of BNKZ should lead to a general increase in polarisation due to the occurrence of O-T and R-T phase coexistence. However, in this study it was found that both P_s and P_r gradually decreased with increasing BNKZ content. This is possibly due to the reduction of grain size, which is known to cause a decrease of remnant polarisation [9]. Furthermore, the core-shell type microstructure could have an additional effect on the polarisation switching characteristics. In contrast, Ochoa et al. [27] found that $(\text{K}_{0.44}\text{Na}_{0.52}\text{Li}_{0.04})(\text{Nb}_{0.86}\text{Ta}_{0.10}\text{Sb}_{0.06})\text{O}_3$ ceramic showed increases in P_s and P_r with increasing temperature due to the O-T phase transformation and reached the highest point at the temperature associated with the O-T mixed phase.

Figure 6. P-E hysteresis loops of 1BNKZ to 5BNKZ ceramics measured at room temperature.

4. Conclusions

Measurements of the temperature-dependent dielectric properties showed that the addition of BNKZ into KNNS leads to a reduction of the orthorhombic to tetragonal

phase transformation temperature, T_{O-T} . Moreover, the relaxor ferroelectric behaviour was observed in KNNS-BNKZ ceramics. By combining the dielectric results with high resolution synchrotron XRD, it was confirmed that the addition of BNKZ into KNNS causes O-T and R-T phase coexistence in 2BNKZ and 4BNKZ, respectively, which is consistent with previous observations [8]. However, the occurrence of core-shell type microstructures in KNNS-BNKZ ceramics due to chemical heterogeneity showed the strong effect on phase formation as well as the functional properties.

5. Acknowledgements

We thank Diamond Light Source for access to beamline I11 (proposal number EE13116) that contributed to the results presented here. The assistance of Dr Sarah Day and Prof Chiu Tang is gratefully acknowledged. Also, JT would like to thank Thammasat University for financial support in the form of a PhD scholarship at the University of Manchester.

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